

1 Introduction

Light hydrocarbons in the marine environment constitute an important part of the carbon cycle and play an important role in global climate change (Kansal, 2009; Exton et al., 2015; Li et al., 2022). Methane (CH₄) is a key greenhouse gas with a much higher global warming potential than CO₂ (IPCC, 2018). Non-methane hydrocarbons (NMHCs), including ethane, ethylene, propane, propylene, and isoprene, can react with NO_x compounds in the atmosphere, and contribute to the formation of atmospheric ozone (Atkinson, 2000). The ocean has been identified as a natural source of climate-active gases, including methane and NMHCs (Borges et al., 2016; Bourtsoukidis et al., 2020; Rosentreter et al., 2021). The global ocean emission rates of CH₄ and NMHCs (C₂-C₄) are estimated to be 10 Tg yr⁻¹ and 2.1 Tg yr⁻¹, respectively (Plass-Dußer et al., 1995; Saunois et al., 2016). Therefore, it is crucial to understand the dynamics and controls of these light hydrocarbons in pelagic marine systems.

The distribution of methane and NMHCs in the upper ocean is controlled by multiple factors. Marine algae are a potential source of hydrocarbons to the marine environment and their production rates are dependent on phytoplankton species (Broadgate et al., 2004; Damm et al., 2008). Additionally, bacteria are also important producers of hydrocarbons. For example, carbon-phosphorus (C-P) substrates (e.g., methylphosphonate and 2-hydroxyethylphosphonate) can be used by marine bacteria to induce methane and ethylene supersaturation in phosphate-depleted water column (Karl et al., 2008; Repeta et al., 2016; Sosa et al., 2020; Mao et al., 2022). In addition, photochemical degradation (Ratte et al., 1998; Riemer et al., 2000; Li et al., 2020) and bacterial oxidation (Steinle et al., 2015) of light hydrocarbons can also influence the balance of dissolved gases.

The Kuroshio and Oyashio currents are two western boundary currents that intersect in the North Pacific where Kuroshio and Oyashio Extension (KOE) are located. Mesoscale eddies frequently occurred in the KOE and their dynamics can influence the local climate and hydrography of the system (Williams et al., 2007; Itoh and Yasuda, 2010; Tozuka et al., 2017; Zhou et al., 2023). The positive/negative sea surface temperature anomalies generally observed in anticyclones/cyclones will influence the sea surface heat and momentum fluxes and further influence the mixed layer depth (MLD, Hausmann et al., 2017; Gaube et al., 2019). Anticyclonic eddy can deepen the MLD by increasing stratification and hindering nutrient upwelling thereby decreasing primary production rates (Shih et al., 2020). In contrast, upwelling in cyclonic eddies can thin the MLD and also transport nutrients vertically from the deeper water to the euphotic zone (Spingys et al., 2021). Furthermore, eddy-induced changes in nutrient profiles and phytoplankton community structure can influence the distribution of dissolved gases (Jickells et al., 2008; Weller et al., 2013; Sugimoto et al., 2017). A previous study demonstrated that a phytoplankton bloom in a southwest Pacific mesoscale eddy was associated with increased CH₄ concentrations in the surface mixed layer (Weller et al., 2013). However, the effect of mesoscale eddies on dissolved gases, including CO₂, methane, and NMHCs in marine waters remain poorly understood. In order to accurately assess global oceanic carbon emission fluxes, it is necessary

to consider the effect of eddies on the climate-relevant gas emission fluxes (Yoshikawa et al., 2014; Li et al., 2019; Farias et al., 2021). As such, we conducted in situ measurements within an anticyclonic eddy in the North Pacific to constrain its influence on CO₂, methane, and NMHCs dynamic in the KOE.

2 Materials and methods

2.1 Mesoscale eddy hydrography

A mesoscale anticyclonic eddy was visited in Northern Pacific on board the R/V DongFangHong 312 during September 8-9, 2019 (Figure 1). This mesoscale eddy formed in early August, separated from the KE in September and dissipated in November (Figure S1). The geostrophic current velocity of the eddy is about 0.47 m s⁻¹, and the eddy radius reached about 41.7 km. These mesoscale parameters were estimated using the sea level anomalies (SLA) data (Faghmous et al., 2015). The distribution of SLA were obtained from the Copernicus Marine Environment Monitoring Service (CMEMS) (<https://marine.copernicus.eu>) at 0.25° resolution. In addition, the data of temperature and salinity from 5-1000 m for the Oyashio and Kuroshio water masses were also downloaded from CEMES for reference (Figure 1C). According to the SLA data, sampling sites were categorized as the eddy outside (sites E1, E2; 0.05-0.07 m) and eddy core (sites E4-E9; 0.50-0.16 m). Shown in the T-S diagram (Figure 1C), the water mass of eddy core retained the high salinity characteristics of the Kuroshio extension. In addition, the sparse isotherms illustrated the strong stratification of the water mass occurred at the upper column in the eddy core (Figure 2). For the eddy outside, the water mass below the MLD is mainly characterized by low temperature and low salinity, with obvious Oyashio characteristic. Furthermore, the eddy core was accompanied by the formation of deeper MLD (24.20-24.02 m) compared to the eddy outside (12.39-13.58 m).

2.2 Sample collection

Water samples above 250 m were collected at sites E1 to E9 during the expedition (Figure 1). The temperature and salinity of seawater were obtained from an SBE 911 plus conductivity/temperature/depth (CTD) probe. The MLD was defined by the threshold value of temperature (0.2°C) from the surface value at 10 m depth (de Boyer Montegut et al., 2004). Seawater samples for CH₄ and NMHCs analysis were collected from 12 L Niskin bottles. Subsamples were gently introduced into 120 mL brown glass bottles through a rubber tube and overflowed 3 times of the bottle volume, and then 50 mL saturated HgCl₂ solution was added to inhibit biological activities. Samples were then sealed with aluminum caps containing Teon septa and temporarily stored at 4°C in the dark (Wu et al., 2021; Li et al., 2022). About 1 L seawater was filtered through Whatman filter (0.7 μm) for chlorophyll a (Chl-a) measurement and filters were frozen at -20°C. For the analysis of single phytoplankton cells, 50 mL seawater at sites E1, E5, and E9 was preserved in acidic Lugol's solution at a volumetric ratio of



1:100. For the nutrients sample, ~60 mL seawater was filtered through 0.45 µm cellulose acetate filters and the filtrates were stored frozen at -20 °C. Dissolved organic carbon (DOC) samples were filtered using precombusted (450 °C for 3 hours) 0.7 µm GFF filters. All DOC samples were preserved in precombusted glass bottles with acid cleaned (10% HCl) and frozen at -20 °C until analysis. Furthermore, air samples for determining atmospheric concentrations of methane and NMHCs from three sites (E1, E4, E9) were collected in 3 L fused-silica lined stainless-steel canister (Restek, USA) from 10 m above the sea surface.

2.3 Biogeochemical analyses

Seawater CH₄ concentration was determined using a cryogenic purge and trap system connected to a gas chromatograph with a flame ionization detector (GC, Agilent 8090, USA). Dissolved CH₄ of seawater was purged with high-purity nitrogen gas (50 mL min⁻¹) and trapped at 1/8 stainless steel pipe (Porapak Q, 80-100 mesh) by liquid nitrogen. After purging for 10 minutes, methane was released from the trap loop by heating in a boiling water bath. The gaseous sample was introduced into gas chromatography equipped with the flame ionization detector and capillary column (HP-PLOT/Q, 30 m × 0.32 mm ID, 0.25 mm). The oven temperature was maintained at 90 °C

C, and the peak of the sample was shown in 2 minutes after sample introduction. The method detection limit for methane analysis in this study was 0.1 nmol L⁻¹, and the standard deviation was less than 3%.

Similarly, seawater NMHCs concentration was measured using a cryogenic purge and trap system connected to a gas chromatography-mass spectrometer (GC-MS, Agilent 7890A/5975C, USA). About 50 mL seawater was injected into the sparging chamber and purged with high-purity helium gas (50 mL min⁻¹) for 15 minutes. Anhydrous magnesium perchlorate (Mg(ClO₄)₂) and sodium hydroxide (NaOH) were used as desiccants to remove moisture. The GC was equipped with an Rt-Alumina Bond/KCl capillary column (30 m × 0.32 mm ID, 0.25 mm), and the temperature program was set as follows: oven temperature started at 40 °C for 6 minutes, gradually raised to 120 °C with the rate of 5 °C min⁻¹, held at 120 °C for 8 minutes, raised to 170 °C at 30 °C min⁻¹ and held at 170 °C for 10 minutes. The detection limit of the method was 0.5 pmol L⁻¹. In addition, atmosphere samples were measured using the GC-MS coupled with an atmospheric pre-concentrator (Nutech 8900DS, USA).

Dissolved oxygen (DO) was measured on board followed by the Winkler titration method (Bryan et al., 1976). The concentration of Chl-a was analyzed with a fluorescence spectrophotometer (Hitachi F-4500, Japan). Flow cytometry was used to determine the single phytoplankton cells (Chen et al., 2017). Pigments were excited with a



488 nm laser beam delivered by a 20 mW solid state laser. Each intercepted cell was characterised by five scatter and fluorescence signals, namely sideward scatter, red fluorescence (FLR; 668/734 nm), orange fluorescence (FLO; 601/668 nm) and yellow fluorescence (FLY, 536/601 nm). The flow cytometer was driven by the CytoSub software and data was analyzed using CytoClus3 software (CytoBuoy). Three sizes of phytoplankton community, namely picophytoplankton, nanophytoplankton and microphytoplankton were determined at sites E1, E5, and E9. The concentration of dissolved inorganic nitrogen (DIN, nitrate plus nitrite), phosphate (PO_4^{3-}), and silicate (SiO_3^{2-}) were determined with an AA3 nutrients analyzer (SEAL Analytical, UK). Samples for DOC were analyzed via the high-temperature catalytic oxidation method (Guo et al., 2011). ~10 mL sample was acidified to pH 2-3 with 2 mol L^{-1} HCl, and bubbled with high-purity nitrogen (N_2) gas to remove dissolved inorganic carbon. Organic carbon was catalytically oxidized to CO_2 (with 0.5% Pt- Al_2O_3) at 680°C , which was measured with a non-dispersive infrared gas analyzer (Shimadzu TOC-VCPH, Shimadzu Co., Japan).

2.4 Underway observation

The underway observation of pCO_2 was carried out from 152.0°E to 156.0°E along the transect of 37.6°N . Seawater at approximately 3 m below the sea surface was pumped into the laboratory and the flow rate of seawater was set to 1 L min^{-1} . After separation of gaseous and aqueous phase using a spraying equilibrator, gaseous samples were filtered through a 0.1 mm filter membrane before analysis using a Gas Concentration Analyzer (Picarro G2131-i, USA) with the cavity ring-down spectroscopy (CRDS). Air samples were collected from 10 m above the sea surface at the flow rate of 300 mL min^{-1} . The instrument was calibrated with standard gas at CO_2 levels of 200 ppmv, 400 ppmv and 600 ppmv every 3 hours (China National Research Center for Certified Reference Materials) with an uncertainty is less than 0.1%. Sea surface temperature (SST), salinity (SSS), and Chl-a were measured by Ferrybox (4H-JENA, Germany), a highly integrated, automatic device equipped with multiple probes (SBE45, USA; Seapoint Chlorophyll Fluorometer, USA).

2.5 Nutrients transport fluxes

Nutrient flux (N_{flux}), representing the vertical diffusive flux of nitrate through the base of the euphotic zone, was calculated using Fick's first law of diffusion as Eq. 1 (Shih et al., 2020).

$$N_{flux} = K_z d_N = d_z \quad (\text{Eq. 1})$$

In Eq. 1, K_z was the average diffusion coefficient in the upper 150 m water column and d_N/d_z was the vertical gradient of nitrate between the MLD and 150 m. The diffusion coefficient was estimated using Eq. 2:

$$K_z = 0.25 \frac{\epsilon}{N^2} \quad (\text{Eq. 2})$$

where ϵ represented the turbulent energy dissipation rate and was determined by the techniques outlined by Dillon (1982) and N was the buoyancy frequency, which was calculated as the vertical density gradient between the MLD and 150 m. In addition, water column inventories of N (I-N), P (I-P), and Si (I-Si) were calculated by the trapezoidal integration for the upper 150 m.

In the interior of the ocean, the biogeochemical cycle of nitrate and phosphate is affected by physical transport and the process of nitrification and denitrification, which will affect the growth of phytoplankton. N^* and P^* as a tracer indicating the perturbation of phytoplankton growth was calculated by Eq. 3 and Eq. 4 based on the Redfield ratio (Gruber and Sarmiento, 1997).

$$N^* = (N - 16P + 2.9) \times 0.87 \quad (\text{Eq. 3})$$

$$P^* = P - N/16 \quad (\text{Eq. 4})$$

In Eq. 3 and Eq. 4, N and P represent the concentration of inorganic nitrogen (nitrite + nitrate + ammonium) and phosphate, respectively. N^* or P^* value near zero reflects nutrient conditions approximately equal to the Redfield ratio, whereas positive and negative values of N^* or P^* were associated with non-conservative behavior.

2.6 Air-sea gas flux calculations

The air-sea fluxes of CO_2 (unit: $\text{g C m}^{-2} \text{y}^{-1}$), methane and NMHCs (unit: $\text{mol m}^{-2} \text{day}^{-1}$) was calculated with following equation (Eq. 5):

$$F = k(C_{sea} - C_{atm}) \quad (\text{Eq. 5})$$

where C_{sea} and C_{atm} were the concentration of hydrocarbons in the surface seawater (unit: mol L^{-1}) and atmosphere (unit: ppm), respectively. k , the gas transfer velocity, calculated by the empirical formula proposed by Wanninkhof (1992) as Eq. 6.

$$k = 0.251u^2 (Sc=660)^{-0.5} \quad (\text{Eq. 6})$$

where Sc was the Schmidt number in seawater, and u was the wind speed at 10 m height (unit: m s^{-1}).

3 Results

3.1 Physical and biogeochemical property

The temperature-salinity (T-S) diagrams of sites E1 and E2 reflect the rapid change of temperature and salinity in the water column above 50 m (Figure 1C). In contrast, the sampling sites at the eddy core generated a T-S diagram representing the more stratified water column, and an obvious feature in the isothermal curve and isohaline curve occurred at the eddy core (Figure 2). Additionally, the SST decreased from 26.7°C to 25.2°C and SSS varied from 34.2 to 34.6 between sites E3 and E7 (Figure 3). Across all sampling sites, the MLD ranged from 10.9 m to 30.1 m with the shallowest at the site E2 and the maximum of MLD at the edge of eddy (site E8) (Table 1).

We measured biogeochemical parameters along with nutrient distributions between the eddy outside and eddy core (Table 1). The distribution of nitrate and phosphate showed a remarkable difference between the eddy outside (nitrate, 8.43–13.37 mmol L^{-1} ; phosphate, 0.41–0.27 mmol L^{-1}) and eddy core (nitrate, 4.03–6.33 mmol L^{-1} ; phosphate, 0.11–0.15 mmol L^{-1}). N_{flux} was higher at the eddy outside (2.46–11 $\text{mmol m}^{-2} \text{d}^{-1}$) than at the eddy core (0.67–0.63 $\text{mmol m}^{-2} \text{d}^{-1}$). Phytoplankton biomass might also vary following the change of nutrient distribution. The range of Chl-*a* at the DCM for the eddy outside and inner were 0.33–1.10 mg L^{-1} and 0.50–1.09 mg L^{-1} , respectively. The deep chlorophyll maximum (DCM) was considerably deepened at site E5 (80 m) corresponding to elevated DOC concentrations below the MLD (Figure 2). DOC concentration below MLD at the eddy core ranged from 0.59 to 1.42 mg L^{-1} with a maximum value at 105 m depth for site E6. Furthermore, *Synechococcus* was the dominant species at site E1, followed by picoeukaryotes, while dominant species switched to *Prochlorococcus* at the eddy core. The growth of microeukaryotes and nanoeukaryotes was largely inhibited and contributed to less than 0.9%–2.4% of phytoplankton community (Figure 4).

3.2 Spatial distributions of methane and NMHCs

The change in dissolved CO_2 was similar to that of the SST, decreasing from 406.1 matm at the eddy outside to 377.5 matm at the eddy core (Figure 3). Across all sites, dissolved CH_4 concentrations ranged from 0.3 nmol L^{-1} to 9.9 nmol L^{-1} , with an average of 3.4–2.2 nmol L^{-1} . The maximum of CH_4 was observed at 80 m at the eddy core (site E05; Figure 5). At the sites E7–E9, CH_4 concentrations were < 4 nmol L^{-1} and did not vary substantially with depth. Average concentrations of ethane, ethylene, propane, and propylene were 33.4–18.3, 62.9–22.8, 14.1–5.6, 33.5–12.3 pmol L^{-1} , respectively. Similar distributions of light alkanes and alkenes (C2–C3) were observed at site E2, and abundant hydrocarbons occurred at the surface layer and 180 m (Figure 5). Higher NMHCs concentrations were observed at the eddy core,

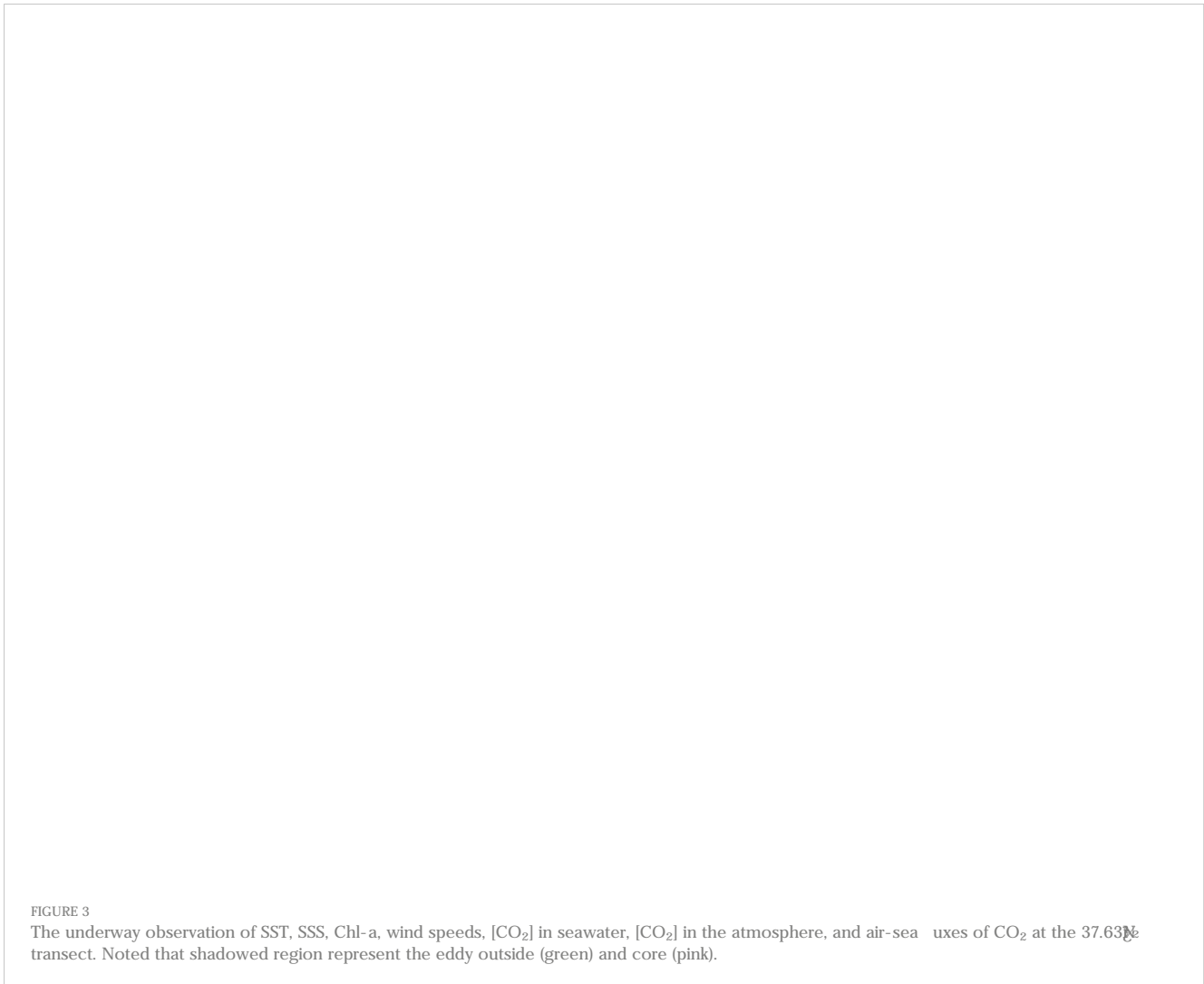


FIGURE 3

The underway observation of SST, SSS, Chl-a, wind speeds, $[\text{CO}_2]$ in seawater, $[\text{CO}_2]$ in the atmosphere, and air-sea fluxes of CO_2 at the 37.63°E transect. Noted that shadowed region represent the eddy outside (green) and core (pink).

with the maximum concentrations observed at site E4. Similar to methane, dissolved ethane, propane, and propene did not vary with depth at the sites E7-E9. In contrast, ethylene concentrations were elevated up to $\sim 85 \text{ pmol L}^{-1}$ at site E7. Isoprene, commonly produced by phytoplankton, ranged from 3.5 pmol L^{-1} to 27.9 pmol L^{-1} , with an average of $10.4 \pm 6.7 \text{ pmol L}^{-1}$. The spatial distribution of isoprene closely matched that of Chl-a concentration with the maximum located above DCM.

3.3 Air-sea fluxes of CO_2 , methane and NMHCs

We compared air-sea fluxes of CO_2 , methane, and NMHCs throughout the eddy to elucidate the influence of the eddy on the distribution of these compounds (Figures 3, 6). Wind speed was elevated at the eddy core sites and ranged from 1.10 to 4.65 m s^{-1} with an average of $3.21 \pm 0.81 \text{ m s}^{-1}$. The maximum wind speed was observed at site E4 (4.65 m s^{-1}). In contrast, wind speed decreased to 0.35 m s^{-1} at the site E9. Air-sea fluxes of CO_2 across the eddy varied from -3.1 to $0.06 \text{ mmol m}^{-2} \text{ d}^{-1}$, with the maximum value calculated at the eddy core (Figure 3). Note that the negative value indicates

that the ocean is a sink for the atmosphere, while the positive value represents that the ocean is a source. The air-sea fluxes of CH_4 ranged from 0.1 - $1.6 \text{ mmol m}^{-2} \text{ d}^{-1}$ and was substantially lower at the sites E8-E9 relative to the other sites (Figure 6). The calculated air-sea fluxes of the five NMHCs varied from 0.9 - 34.7 (ethane), 2.3 - 108.3 (ethylene), 0.5 - 14.0 (propane), 1.2 - 34.7 (propylene), and 0.4 - $5.8 \text{ nmol m}^{-2} \text{ d}^{-1}$ (isoprene), respectively. Except for isoprene, the maximum air-sea fluxes of NMHCs occurred at site E2. Instead, the maximum isoprene ($4.2 \pm 2.2 \text{ nmol m}^{-2} \text{ d}^{-1}$) fluxes were observed at the eddy core (site E5).

4 Discussion

4.1 Nutrients dynamic influenced by anticyclonic eddy

Eddy and other mesoscale processes (lateral advection, eddy pumping and eddy-driven stratification) are prevalent in the ocean and influence the distributions of nutrients and organic carbon in the upper water column (McGillicuddy, 2016; Shih et al., 2020). The maximum value of N fluxes at the eddy outside manifests the

TABLE 1 Summary of environmental variables, including the mixed layer depth (MLD), temperature, and salinity at MLD; the maximum value of Chl-a and its depth (DCM); the average concentration of nitrate, nitrite, DIN, phosphate, silicate, dissolved organic carbon (DOC), and dissolved oxygen (DO), respectively; the abnormal of Red field ratio for N* and P*; water column inventories (0–150 m) of DIN, silicate, phosphate, and Chl-a for I-N, I-Si, I-P, and I-Chl-a; the calculated values of diffusion coefficient (K_z); the vertical flux of N; the average concentration of methane, ethane, ethylene, propane, propylene, and isoprene, respectively.

Parameters	Eddy Outside		Eddy Core					
	E1	E2	E4	E5	E6	E7	E8	E9
MLD (m)	13.6	12.4	23.9	25.7	18.1	27.0	30.1	20.3
Temperature _{MLD} (°C)	24.9	25.5	24.2	24.7	24.2	24.1	24.2	24.3
Salinity _{MLD}	34.0	34.1	34.5	34.4	34.5	34.5	34.5	34.4
DCM (m)	80	55	55	105	25	55	65	55
Chl-a _{DCM} (mg L ⁻¹)	0.23	0.42	0.46	0.46	0.58	0.57	0.33	0.57
Nitrate (mmol L ⁻¹)	8.77 ± 0.57	8.12 ± 0.16	3.39 ± 0.07	3.21 ± 0.53	2.85 ± 0.88	3.46 ± 0.84	4.27 ± 0.85	5.61 ± 0.77
Nitrite (mmol L ⁻¹)	0.04 ± 0.01	0.08 ± 0.04	0.07 ± 0.05	0.08 ± 0.05	0.08 ± 0.05	0.06 ± 0.06	0.04 ± 0.03	0.05 ± 0.06
DIN (mmol L ⁻¹)	9.20 ± 0.57	8.38 ± 0.06	4.38 ± 0.06	4.67 ± 0.85	3.50 ± 0.54	5.65 ± 0.95	5.98 ± 0.72	6.70 ± 0.77
Phosphate (mmol L ⁻¹)	0.44 ± 0.27	0.39 ± 0.26	0.08 ± 0.09	0.05 ± 0.06	0.09 ± 0.12	0.05 ± 0.06	0.12 ± 0.19	0.20 ± 0.21
Silicate (mmol L ⁻¹)	16.5 ± 0.9	13.8 ± 0.9	5.4 ± 0.9	4.8 ± 0.7	6.45 ± 0.89	5.86 ± 0.54	7.72 ± 0.24	8.45 ± 0.75
DOC (mg L ⁻¹)	0.82 ± 0.09	1.07 ± 0.77	0.95 ± 0.25	0.83 ± 0.10	0.94 ± 0.21	0.74 ± 0.05	0.76 ± 0.05	0.75 ± 0.09
DO (mg L ⁻¹)	8.33 ± 0.63	9.27 ± 0.86	8.78 ± 0.58	8.36 ± 0.41	10.70 ± 0.45	10.60 ± 0.65	10.46 ± 0.55	10.14 ± 0.67
N* (mmol L ⁻¹)	4.45 ± 0.59	4.43 ± 0.09	5.23 ± 0.71	5.89 ± 0.21	4.26 ± 0.29	6.81 ± 0.87	6.05 ± 0.68	5.62 ± 0.21
P* (mmol L ⁻¹)	-0.14 ± 0.11	-0.14 ± 0.08	-0.19 ± 0.12	-0.24 ± 0.16	-0.12 ± 0.18	-0.31 ± 0.21	-0.25 ± 0.19	-0.22 ± 0.16
I-N (mmol m ⁻²)	1227	1013	514	580	394	728	614	821
I-Si (mmol m ⁻²)	2118	1741	666	650	594	754	755	978
I-P (mmol m ⁻²)	56	45	5.0	6.7	9.8	3.1	5.9	13
I-Chl-a (mg L ⁻¹)	16.4	18.9	24.2	30.4	32.8	16.3	24.5	25.0
K_z (m ² d ⁻¹)	7.61	5.59	0.57	6.00	3.28	0.57	4.37	1.10
N mmol m ⁻² d ⁻¹	2.35	2.57	0.15	1.40	0.52	-0.09	1.61	0.40
CH ₄ (nmol L ⁻¹)	5.0 ± 0.5	2.1 ± 0.2	3.4 ± 0.5	4.7 ± 0.3	0.2	2.2 ± 0.0	2.4 ± 0.3	1.7 ± 0.1
Ethane (pmol L ⁻¹)	56.4 ± 0.7	32.4 ± 2.3	39.4 ± 0.2	29.4 ± 0.7	24.7 ± 0.5	28.9 ± 0.1	25.1 ± 0.6	28.4 ± 0.8
Ethylene (pmol L ⁻¹)	70.8 ± 0.2	73.3 ± 0.1	69.1 ± 0.4	60.2 ± 0.4	52.9 ± 0.2	64.7 ± 0.3	53.1 ± 0.8	59.4 ± 0.3
Propane (pmol L ⁻¹)	10.7 ± 0.38	13.3 ± 0.8	22.4 ± 0.6	18.9 ± 0.2	9.5 ± 0.2	10.4 ± 0.7	13.8 ± 0.4	12.0 ± 0.7
Propylene (pmol L ⁻¹)	32.3 ± 0.89	35.4 ± 0.9	42.3 ± 0.0	32.7 ± 0.9	29.9 ± 0.0	34.7 ± 0.3	28.3 ± 0.0	32.2 ± 0.0
Isoprene (pmol L ⁻¹)	11.8 ± 0.9	10.9 ± 0.0	16.0 ± 0.3	10.6 ± 0.3	7.9 ± 0.7	8.0 ± 0.9	9.4 ± 0.9	7.9 ± 0.3

nutrient supplement of the deep water (Table 1), which is consistent with previous studies of nutrient enrichment at the edge of anticyclonic eddies (Zhou et al., 2013; McGillicuddy, 2016). Likewise, the value of I-N, I-P and I-Si were higher at the eddy outside compared to other sites in the eddy system (Table 1). Therefore, the upper water column of the eddy core represents a nutrient limitation compared to the eddy outside.

Furthermore, N* increased from 4.44 ± 0.35 mmol kg⁻¹ (eddy outside) to 5.71 ± 0.35 mmol kg⁻¹ (eddy core). Phosphate concentrations were limited as reflected by a decrease in P* (-0.14 ~ -0.26). The limitation of nutrients will restrict the development of phytoplankton in the anticyclonic eddy. In our study, large size eukaryotes such as nanophytoplankton and microphytoplankton

were detected with low abundance, but Prochlorococcus was highly abundant at the eddy core. In addition, elevated phytoplankton cell mortality rates and cell lysis rates at the anticyclonic eddy could be responsible for the higher DOC production (Lasternas et al., 2013). Taken together, the process of anticyclonic eddy leads to a change in nutrient conditions, phytoplankton structure in the upper ocean.

4.2 Potential controlling factors of methane and NMHCs in the eddy

Methane production is directly linked to the N, P and C cycles as feedback of environmental perturbations such as changes in

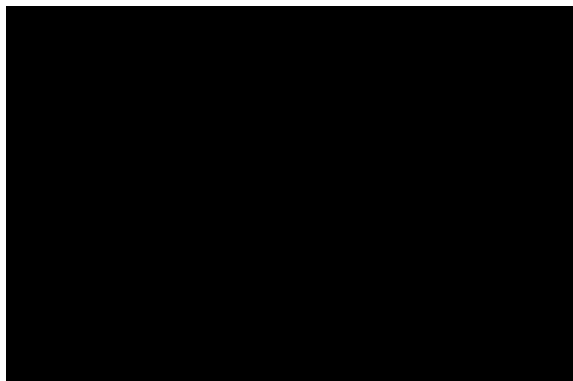


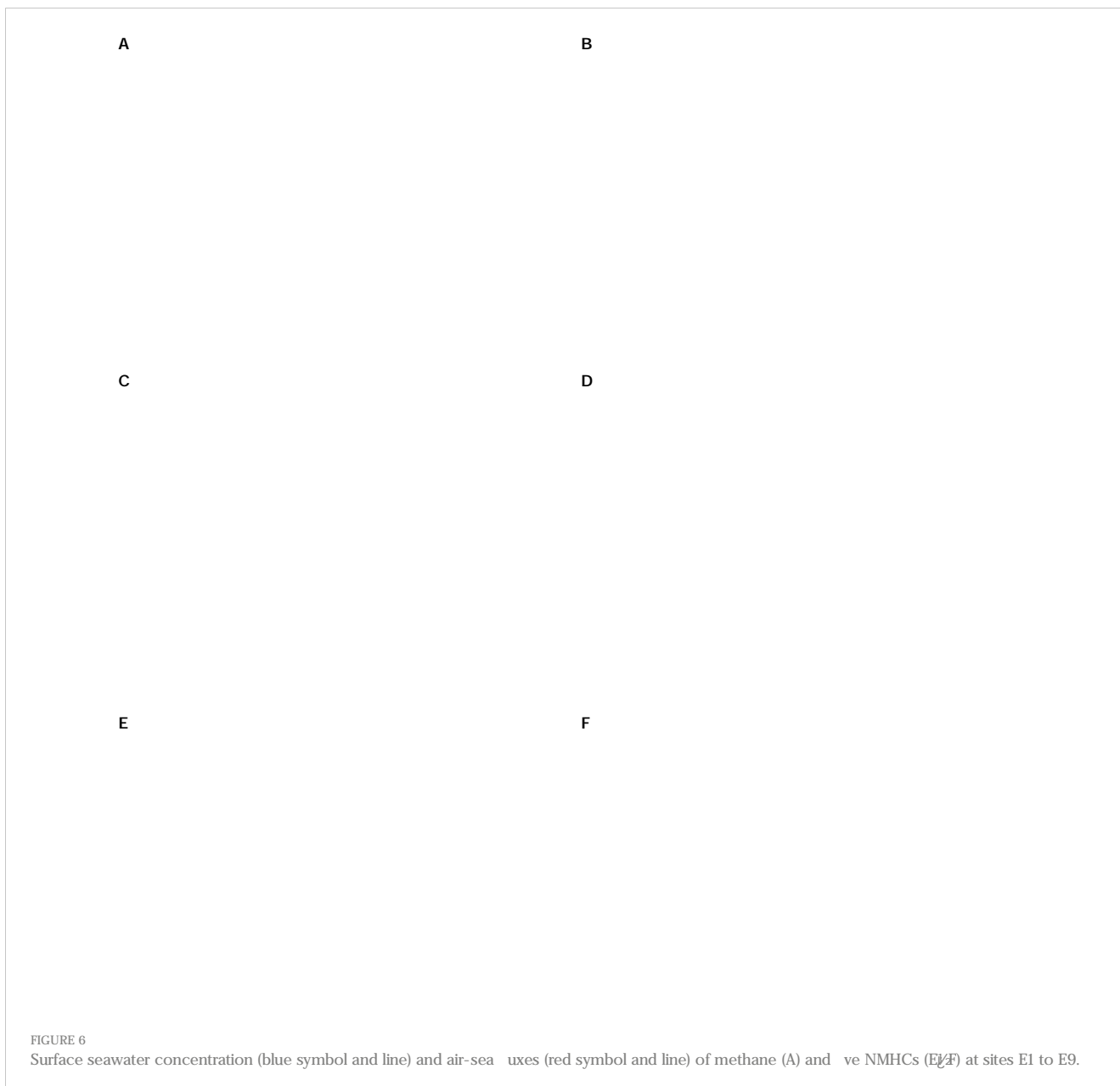
FIGURE 4
The proportion of the phytoplankton community in the DCM layer at the eddy outside (site E1), core (sites E5, E9).

phytoplankton structure and nutrient supply (Damm et al., 2008; Karl et al., 2008). The methane maximum at site E5 in the eddy outside may be related to the low nutrient level in the water column (Weller et al., 2013). Under nutrient limited conditions, microorganisms can use compounds such as methylamine and methylphosphonate (MPn) as alternative sources of P and N (Repeta et al., 2016; Ye et al., 2020; Mao et al., 2022). The microbial utilization of MPn lead to the lyase of C-P and CH₄ could be produced as a by-product (Repeta et al., 2016; Acker et al., 2022). Furthermore, most semi-labile DOM including MPn is produced by Prochlorococcus in the open ocean (Repeta et al., 2016). Hence, the growth of Prochlorococcus at the eddy core facilitated methane production through the C-P pathway. The vertical distribution of methane is homogenous at sites E7 and E8, which could be caused by the increasing depth of the mixing layer.

Isoprene is produced biologically by phytoplankton, and its distribution reflects the abundance and structure of phytoplankton (Broadgate et al., 2004; Li et al., 2019; Conte et al., 2020). As such, a subsurface maximum was observed for the concentration of isoprene in the water column, generally consistent with the depth of DCM (Figure 5). Interestingly, strong correlation between the isoprene and Chl-a was observed at the eddy core ($R^2=0.38$; $n=32$; $P<0.01$), while little correlation occurred at the eddy outside ($R^2=0.003$; $n=11$; $P>0.01$). Such difference indicated that phytoplankton structure could be important in the isoprene distribution as the production of isoprene was species-dependent (Broadgate et al., 1997; Kurihara et al., 2010; Li et al., 2021). It has been demonstrated that Prochlorococcus was an important producer of isoprene in the open ocean (Shaw et al., 2003). At the eddy core, the abundance of Prochlorococcus was much higher than other species, suggesting that Prochlorococcus dominate the production of isoprene and also explained the observed correlation between isoprene and Chl-a. Previous laboratory studies have found various species of microeukaryotes, nanoeukaryotes, picoeukaryotes and cyanobacteria (including Prochlorococcus and Synechococcus) are all the isoprene producers (McKay et al., 1996; Shaw et al., 2003). Although the size of the phytoplankton cell

FIGURE 5
The vertical distribution of methane and five NMHCs (ethane, propane, ethylene, propylene, and isoprene) in the anticyclonic eddy. Note that the blank data points without color filled indicated the depth of DCM.

influences the production rate, the abundance of Prochlorococcus is one or two orders of magnitude higher than that of other eukaryotes. Therefore, the succession of small-size phytoplankton leads to the variation of isoprene in the anticyclonic eddy. In contrast to isoprene, other NMHCs have different sources, and the distribution could be impacted by other environmental factors. Previous studies found light NMHCs can be produced through the degradation of polyunsaturated lipids, which originate from marine phytoplankton (Lee and Baker, 1992; Broadgate et al., 2004; Plettner et al., 2005). The concentrations of ethane and ethylene at the eddy outside were higher than those at the eddy core, in which the abundant DOC was enriched at the surface layer. The maximum of NMHCs occurred near DCM at the sites E5 and E6, which could be caused by the degradation of phytoplankton-related organic matter. Similar to methane, ethylene and propylene can also be produced by



the bacterial degradation of phosphonates (Repeta et al., 2016). Increased phytoplankton mortality during the eddy process would enhance the autolysis of phytoplankton cells, which could contribute to the subsurface maximum for ethane and propane (McKay et al., 1996). Principal component analysis (PCA) is used to analyze the influence of environmental factors on methane and NMHCs in eddy process (Figure 7). The results of PCA analysis illustrate that two principal components explain >60% of the total variation. For Figures 7A, B, PC1 is related to the temperature and nutrients (DIN, phosphate, silicate) and PC2 is significantly loaded by salinity, DOC. Methane exerts major contribution to PC2, indicating that the spatial distribution of DOC influenced by eddy activity affects the methane production, such as MPn pathway. By

contrast, isoprene was strongly loaded in the PC2, illustrating that isoprene may be more sensitive to temperature. Previous studies with phytoplankton monocultures have found positive dependence of isoprene production rate on temperature within $-0.8 \sim 23^{\circ}\text{C}$, which is restricted to the enzymatic activities (Shaw et al., 2003; Simo et al., 2022). Our results also indicated 23°C is the optimum temperature for isoprene production ($R^2=0.286$; $n=34$; $P<0.01$; Figure S2), which is responsible for the maximum value at the subsurface (Simo et al., 2022). In addition, the response of light alkanes and alkenes to environmental factors may be complicated, especially the biological factors are relatively weak (Figure 7C). Furthermore, the strong correlation was observed between light alkanes and alkenes, which suggested that similar source or removal



processes was affected by eddy activity (Figure 7D).

4.3 Effect of anticyclone eddy on the air-sea fluxes of gases

The occurrence of ocean eddies exerts crucial influence on the vertical structure of the marine atmospheric boundary layer and further affects many atmospheric processes including air turbulence, wind speeds, as well as air-sea exchange (Frenger et al., 2013; Weller et al., 2013; Pezzi et al., 2021). Apparently, the sites E2-E7 were influenced by a strong wind field ($2.93\text{--}3.47\text{ m s}^{-1}$), while the wind speed at the sites E1 (1.88 m s^{-1}), E8 (0.97 m s^{-1}) and E9 (1.11 m s^{-1}) were relatively calm. For air-sea exchange, the calculated air-sea fluxes of CO_2 in this anticyclone eddy (-1.04 to $0.80\text{ mmol m}^{-2}\text{ d}^{-1}$) were lower than those reported values in the KOE region ($-6.5\text{ mmol m}^{-2}\text{ d}^{-1}$, Sutton et al., 2017; -2.70 to 2.31

$\text{mmol m}^{-2}\text{ d}^{-1}$, Yan et al., 2023) and North Pacific extratropic (2.42 to $0.67\text{ mmol m}^{-2}\text{ d}^{-1}$, Ishii et al., 2014). In addition, the sink of CO_2 at the eddy core (-1.27 to $0.78\text{ mmol m}^{-2}\text{ d}^{-1}$) was stronger than the eddy outside (-0.60 to $0.57\text{ mmol m}^{-2}\text{ d}^{-1}$). According to Eq. 5 and Eq. 6, the elevated air-sea CO_2 exchange at the eddy core could be joint result of lower SST, higher pCO_2 , and strong winds. Furthermore, air-sea fluxes for methane ($0.10\text{--}1.64\text{ mmol m}^{-2}\text{ d}^{-1}$) and isoprene ($0.42\text{--}5.79\text{ nmol m}^{-2}\text{ d}^{-1}$) were lower than the reported values in the Pacific (methane: $1.64\text{--}2.93\text{ mmol m}^{-2}\text{ d}^{-1}$, Rehder and Suess, 2001; Isoprene: $2.1\text{--}300\text{ nmol m}^{-2}\text{ d}^{-1}$, Matsunaga et al., 2002; Li et al., 2019). Higher methane emission occurred at sites E2 and E5, and the maximum flux of isoprene occurred at site E5 where had the highest surface concentration. For other NMHCs, the emissions of ethane, ethylene, and propylene appear to be attenuated at the site E6-E9 due to the decrease in surface concentration. Collectively, eddy processes can affect local regional sea-air exchange processes by influencing water temperature, gas concentrations and wind

elds. Globally, the effect of eddy on gas emissions could be important given the prevalence of mesoscale eddies in the ocean, which should be considered in the estimation of global sea-air fluxes in the future.

5 Summary

In this study, we investigate the distribution and emission of climate-relevant gases including CO₂, methane, and NMHCs in an anticyclone eddy in the KOE during September 8-9, 2019. The eddy exhibited remarkable nutrient limitation due to the downward isopycnal displacement, which favored the growth of small-size phytoplankton, especially *Prochlorococcus*. Phosphorus limitation within the eddy facilitated the production of methane from the C-P pathway. Significant correlation was observed between dissolved isoprene and Chl-*a* due to the dominance of *Prochlorococcus* at the eddy core sites. The elevated concentrations of ethane, propane, ethylene, and propylene in the water column could be related to the production of DOC. Air-sea fluxes of gases were largely influenced by the anticyclone mesoscale eddy. The ventilation of CO₂ and isoprene increased at the eddy core, while the air-sea fluxes of methane and light NMHCs were lower at the eddy outside due to the reduced wind speeds. Our results indicate mesoscale eddy exerts an important influence on the distribution and emission of light hydrocarbons, short-lived ocean events such as mesoscale eddies should be considered in the future estimates of gas fluxes.

Data availability statement

The original contributions presented in the study are included in the article/Supplementary Material. Further inquiries can be directed to the corresponding authors.

Author contributions

H-HZ and G-CZ designed the study. X-JL performed the experiments, with assistance from JW, and HQ. JW provided NMHCs data and HQ provided DO data. Phytoplankton data was provided by SZ. X-JL organized and analyzed the database, wrote the manuscript and prepared the tables and figures. H-HZ, AM, G-CZ, Z-HC, and R-CZ provided comments on data analysis

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Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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Supplementary material

The Supplementary Material for this article can be found online at: <https://www.frontiersin.org/articles/10.3389/fmars.2023.1181896/full#supplementary-material>

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